

fac-Tris(dimethylformamide)tris(isothiocyanato)iron(III)

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Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean $\sigma(\text{N}-\text{C}) = 0.003\text{ \AA}$

R factor = 0.062

wR factor = 0.146

Data-to-parameter ratio = 16.7

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The reaction of FeCl_3 , NaSCN and $\text{Gd}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ in 1:10 DMF/ H_2O led to the formation of the title compound, $[\text{Fe}(\text{NCS})_3(\text{C}_4\text{H}_7\text{N})_3]$, in which the six-coordinated Fe atom is in an octahedral *fac* environment.

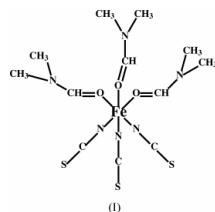
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Comment

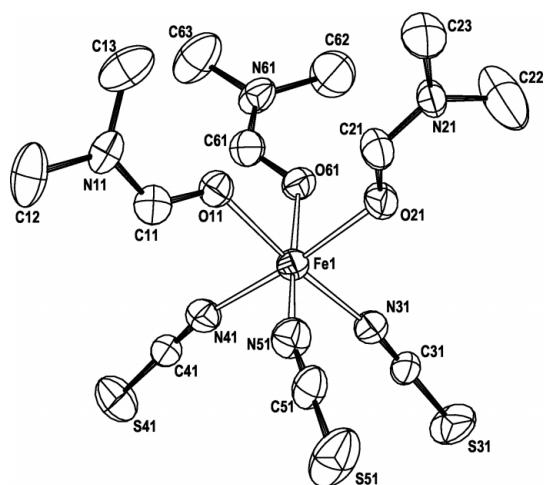
The SCN^- ligand has been extensively investigated in recent years (Vinogradova *et al.*, 2001; Ren *et al.*, 2001; Kuang *et al.*, 2001; Sudbrake & Vahrenkamp, 2001). In most of the reported iron(III) complexes, the SCN^- ligand acts in a monodentate manner (Marchivie *et al.*, 2002; Zhu *et al.*, 2002; Marlin *et al.*, 2002). During our research on 3d–4f metal complexes, we obtained the title compound, (I), a mononuclear iron(III) complex containing the SCN^- ligand. Here we report the structure of (I).



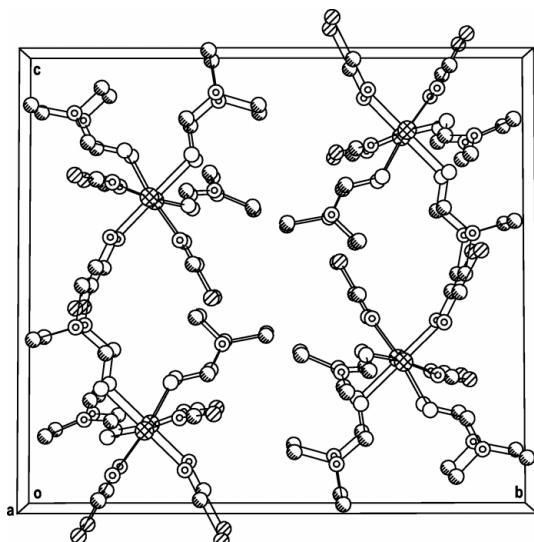
In the title complex (Fig. 1), the six-coordinated Fe^{III} atom is surrounded by three SCN^- anions and three DMF molecules with mean $\text{Fe}-\text{N}$ and $\text{Fe}-\text{O}$ distances of 2.009 (6) and 2.037 (8) Å, respectively. Atoms N41, N51, O61 and O21 are coplanar, and the deviation of atom Fe1 from the above plane is 0.0554 (3) Å. The structure of (I) is similar to that of *fac*-tris(isothiocyanato)tris(triphenylphosphineoxide)iron(III) (Olejnik *et al.*, 1995). All six *fac*-positioned ligands are monodentate. Three near-linear SCN^- ligands in (I) have an average $\text{S}-\text{C}-\text{N}$ bond angle of 179.2 (2)°. The $\text{S}-\text{C}$ and $\text{C}-\text{N}$ bond distances in the SCN^- ligand are comparable to reported values (Olejnik *et al.*, 1995). The packing of (I) in the unit cell is shown in Fig. 2. The crystal structure is stabilized by van der Waals interactions.

Experimental

FeCl_3 (1 mmol, 162 mg) was added to a 10 ml aqueous solution of NaSCN (6 mmol, 486 mg). A clear red solution was formed; $\text{Gd}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ (1 mmol, 451 mg) and DMF (1 ml) were then added successively. The final solution was filtered and allowed to stand in an ambient environment. Black-red prismatic crystals formed after several days.

**Figure 1**

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level and H atoms have been omitted for clarity.

**Figure 2**

The molecular packing, viewed down the *a* axis. H atoms have been omitted for clarity.

Crystal data

$[\text{Fe}(\text{NCS})_3(\text{C}_4\text{H}_7\text{N})_3]$
 $M_r = 449.38$
Monoclinic, $P2_1/c$
 $a = 8.7570 (4)$ Å
 $b = 16.8849 (8)$ Å
 $c = 15.3070 (6)$ Å
 $\beta = 97.591 (2)^\circ$
 $V = 2243.47 (17)$ Å³
 $Z = 4$

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.772$, $T_{\max} = 0.839$
7394 measured reflections

$D_x = 1.330$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 1832 reflections
 $\theta = 1.8\text{--}25.1^\circ$
 $\mu = 0.97$ mm⁻¹
 $T = 293 (2)$ K
Prism, black-red
0.26 × 0.22 × 0.18 mm

3864 independent reflections
1904 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.055$
 $\theta_{\max} = 25.1^\circ$
 $h = -9 \rightarrow 7$
 $k = -20 \rightarrow 14$
 $l = -16 \rightarrow 18$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.146$
 $S = 1.00$
3864 reflections
232 parameters
H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0426P)^2 + 1.9621P]$$

where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\max} = 0.003$
 $\Delta\rho_{\max} = 0.47$ e Å⁻³
 $\Delta\rho_{\min} = -0.32$ e Å⁻³

Table 1
Selected geometric parameters (Å, °).

Fe1–N51	2.000 (2)	S31–C31	1.594 (3)
Fe1–N41	2.012 (2)	S41–C41	1.607 (3)
Fe1–N31	2.015 (2)	S51–C51	1.598 (3)
Fe1–O21	2.0261 (18)	C31–N31	1.146 (3)
Fe1–O61	2.0353 (17)	C41–N41	1.147 (3)
Fe1–O11	2.0481 (18)	C51–N51	1.164 (4)
N51–Fe1–N41	94.47 (9)	O21–Fe1–O61	83.17 (7)
N51–Fe1–N31	93.98 (9)	N51–Fe1–O11	91.21 (8)
N41–Fe1–N31	93.07 (9)	N41–Fe1–O11	89.79 (8)
N51–Fe1–O21	91.02 (9)	N31–Fe1–O11	173.86 (8)
N41–Fe1–O21	173.63 (8)	O21–Fe1–O11	86.87 (7)
N31–Fe1–O21	89.75 (8)	O61–Fe1–O11	85.40 (7)
N51–Fe1–O61	173.41 (9)	N31–C31–S31	179.5 (3)
N41–Fe1–O61	91.17 (8)	N41–C41–S41	178.9 (3)
N31–Fe1–O61	89.11 (8)	N51–C51–S51	179.2 (3)
C31–N31–Fe1	169.9 (2)	C11–O11–Fe1	127.58 (18)
C41–N41–Fe1	165.1 (2)	C21–O21–Fe1	127.97 (18)
C51–C51–Fe1	162.3 (2)	C61–O61–Fe1	129.21 (18)

H atoms were positioned geometrically and allowed to ride on their parent atoms. A rotating group model was used for the methyl groups. The low ratio (0.49) of observed to unique reflections and the relatively large U_{eq} values for the non-H atoms reflect the poor diffraction quality of the crystal.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1994); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Siemens, 1994); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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